

## MOLECULAR GEOMETRY WORKSHEET

Name: \_\_\_\_\_

Period: \_\_\_\_

In the spaces provided draw the electron dot structure and then determine the number of bonding pairs and lone pairs of electrons on the central atom.

	Lewis Structure	Electron Pair Geometry	Bond Angles	Hybridization	Molecular Shape
SeH <sub>2</sub>					
InBr <sub>3</sub>					
MgI <sub>2</sub>					
SiBr <sub>4</sub>					
ICl <sub>3</sub>					
I <sub>3</sub> <sup>-1</sup>					
SeCl <sub>6</sub>					

$\text{RnCl}_4$					
$\text{AsH}_3$					
$\text{SO}_3$					
$\text{BeF}_2$					
$\text{CHCl}_3$					
$\text{CH}_2\text{Cl}_2$					
$\text{PO}_4^{-3}$					
$\text{NO}_3^{-1}$					